

$$Data = \frac{130}{33} + 1 = 4.94$$

$$SP = 1.2$$

$$DF = 1.0$$

$$TED = 5.2 \text{ g/L}$$

$$w = \frac{Data - SP}{Data - DF} = \frac{P_{N_2+He}^{loop}}{P_{N_2+He}^{DIL}} = \frac{3.74}{3.94} = 0.95$$

$$F_{O_2}^{DIL} = \frac{DF}{Data} = \frac{1.0}{4.94} = 0.20$$

Constrain  $P_{N_2} = 2.70$  and check resultant gas density:

$$F_{N_2}^{DIL} = \frac{P_{N_2}}{w \cdot Data} = \frac{2.70}{0.95 \times 4.94} = 0.575$$

$$\therefore F_{He}^{DIL} = 1 - F_{O_2}^{DIL} - F_{N_2}^{DIL} = 1 - 0.20 - 0.575 = 0.225$$

Loop Gas Fractions:

$$F_{O_2}^{loop} = \frac{1.2}{4.94} = 0.243 \quad F_{N_2}^{loop} = w F_{N_2}^{DIL} = 0.95 \times 0.575 = 0.546$$

$$F_{He}^{loop} = w F_{He}^{DIL} = 0.95 \times 0.225 = 0.213$$

Loop Gas Density =

$$(F_{O_2}^{loop} \cdot d_{O_2} + F_{N_2}^{loop} \cdot d_{N_2} + F_{He}^{loop} \cdot d_{He}) \times Data$$

$$= (0.243 \times 1.428 + 0.546 \times 1.251 + 0.213 \times 0.179) \times 4.94$$

$$= 1.068 \times 4.94 = 5.28$$

Resultant gas density is  $> 5.2 \text{ g/L}$   $\therefore$  use gas density formulation:

$$\begin{aligned}
 F_{He}^{loop} &= \frac{Sp \, d_{O_2} + (Data - Sp) \, d_{N_2} - TGD}{Data \, (d_{N_2} - d_{He})} \\
 &= \frac{1.2 \times 1.428 + 3.74 \times 1.251 - 5.2}{4.94 \times 1.072} \\
 &= \frac{1.192}{5.296} = 0.225
 \end{aligned}$$

$$\therefore F_{He}^{DIL} = \frac{F_{He}^{loop}}{w} = \frac{0.225}{0.95} = 0.237$$

$\therefore$  Diluent mix is **TMX 20/24**